SYNTHESIS AND ANTIMICROBIAL EVALUATION OF 3-ISOBUTYL-2,6-DI(THIOPHENYL-3) PIPERIDIN-4-ONE DERIVATIVES

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ABSTRACT

In a wide program of extensive search for new and efficient antimicrobial agents, 3-isobutyl-2,6-di(thiophenyl-3)piperidin-4-one(1) and its derivatives (2-4) were synthesized and characterized by IR, LC-MS, ^1H -NMR, ^{13}C -NMR. The synthesized compounds 1-5 were tested for their antibacterial activity in vitro against streptococcus, bacillus subtilis, Escherichia coli, pseudomonas, and Klebsiella pneumoniae. as the standard drug "streptomycin" was used.

Keywords: 3-isobutyl-2,6-di(thiophenyl-3) piperidin-4-one, ^1H , ^{13}C -NMR spectra, Antibacterial activity, E.coli

1. INTRODUCTION

Piperidine heterocycles play an important role in the field of medicinal chemistry many 2,6-disubstituted derivatives of this class have been found to possess useful biological activities such as antiviral, antitumor [1,2], local anesthetic [3], herbicidal, insecticidal, fungicidal, bactericidal, anti-inflammatory, anti-histaminic, hypotensive, anticancer, central nervous system stimulant and depressant and nerve activities and so forth [4,5,6,7].

The chemistry of hydrazine derivatives such as thiosemicarbazide and its hydrazones is of immense interest owing to their wide synthetic and analytical applications and biological activities [8] which leads to a disruption of the pituitary-thyroid hormonal regulatory system: effective antidotes to paraquat toxicity derivative piperidine are also biologically important and act as neurokinin receptor antagonists and analgesic and anti-

hypertensive agents [9]. Aiming at extending our knowledge in structure activity relationship, we considered that it is valuable to synthesize a system which unites biolabile functional groups like thiosemicarbazide and hydroxylamine into 3-Isobutyl-2,6-di(thiophen-3-yl) piperidin-4-one nucleus.

2. EXPERIMENTAL PART

2.1 MATERIALS

Thin-layer chromatography (TLC) was performed to ascertain proof of the reaction and purity of products. melting points were recorded on a Stuart instrument, IR Spectrometer were recorded on Jusco 300 FT-IR Spectrometer Mass spectra were recorded on a VG 11-250 data acquisition, ^1H , ^{13}C -NMR Spectra were recorded on Bruker 400MHz spectrometer. samples were prepared by dissolving 10mg of substance in 0.5 ml of DMSO or CDCl_3 containing 1% trimethyl silane (TMS). Unless otherwise stated, all the starting materials and reagents

were of high grade ,purchased from Aldrich ,Fluka and Merck .All the solvents were distilled prior to use.

2.2 SYNTHESIS: SYNTHESIS OF 3-ISOBUTYL-2,6-DI(THIOPHENYL-3) PIPERIDIN-4-ONE AND ITS DERIVATIVES:

2.2.1 3-Isobutyl-2,6-di(thiophen-3-yl) piperidin-4-one: (1)

A mixture of ammonium acetate (0.5 mol),thiophene-2-carbaldehyde (1mol) and 5-methylhexan-2-one (0.5 mol) in distilled ethanol was first heated to boiling .after cooling ,the viscous liquid obtained was dissolved in ether (200ml)and shaken with 10 ml of concentrated hydrochloric acid. The precipitated hydrochloride of 3-isobutyl-2,6-di(thiophenyl-3) piperidin-4-one was removed by filtration and washed first with 40 ml mixture of ethanol and ether (1:1) and then with ether to remove most of the coloured impurities . The base was liberated from an alcoholic solution by addition of aqueous ammonia and then diluted with water . It was recrystallized from ethanol .[10]

2.2.2 3-Isobutyl-4-oxo-2, 6-di (thiophen-3-yl) piperidin-1-ium2,4, 6-trinitroph-enolate: (2)

(0.01mol)of 3-Isobutyl-2,6-di(thiophen-3-yl)piperidin-4-one dissolved in 5 ml of ethanol was mixed with (0.01mol)of picric acid dissolved in 5 ml of Ethanol. This mixture was shaken well. The yellow needles formed were recrystallized in benzene [11]

2.2.3 3-Isobutyl-2,6-di(thiophen-3-yl) piperidin-4-one oxime: (3)

About (0.05mol)of 3-Isobutyl-2,6-di(thiophen-3-yl) piperidin-4-one and sodium acetate trihydrate (0.15mol) were dissolved in boiling ethanol and hydroxylamine hydrochloride (0.06 mol) was added. The mixture was heated stirred for 3-4 h then allowed to stand overnight, and poured into

crushed ice. The separated solid was filtered off and recrystallized from ethanol .[12]

2.2.4 3-Isobutyl-2,6-di(thiophe-3-yl) piperidin-4-onethiosemicarbazone: (4)

(0.01mol)of3-Isobutyl-2,6-di(thiophen-3-yl) piperidin-4-one dissolved in boiling ethanol (20ml),2-3drops of con .HCl was added (excess of acid is bad to the reaction) then thiosemicarbazide (0.34g,3.77mmol) was added and the reaction mixture was refluxed for 3 hours on a water bath .The progress of reaction was monitored by TLC .After the complation of reaction ,the reaction mixture was brought to room temperature and the mixture was slowly poured into ice –cold water(400ml) with constant stirring ,the crude product was precipitated. The compound was recrystallized in hot methanol

2.2.5 3-Isobutyl-2,6-di(thiophe-3-yl) piperidin-4-one semicarbazone: (5)

(0.01mol)of 3-Isobutyl-2,6-di(thiophen-3-yl) piperidin-4-one dissolved in boiling ethanol (20ml). Followed by addition of semicarbazidehydrochloride(0.34g,3.77mmol). The contents were refluxed for 3 hours on a water bath .The progress of reaction was monitored by TLC .After the completion of reaction ,the contents were brought to room temperature , the separated solids were filtered .After cooling the mixture was slowly poured into ice – cold water(400ml) with constant stirring .The crude product was precipitated. The compound was Recrystallized in hot methanol [13]

3. RESULTS AND DISCUSSION

The synthesized compound are illustrated in Schem 1. 3-Isobutyl-2,6-di (thiophenyl-3) piperidin-4-one 1 was obtained by the condensation of thiophen-2- carbaldehyde, ammonium acetate and 5-methyl-hexanone-2 in a ratio of 2:1:1. The piperidinim picarte 2 was prepared by mixing equimolar solution of

piperidin-4-one 1 with picric acid in ethanol and stirring the solution for 30 min. Isobutyl-2,6-di(thiophen-3-yl)piperidin-4-one oxime3 was synthesized by treatment of 3-isobutyl-2,6-di(thiophen-3-yl)piperidin-4-one 1 with hydroxylamine hydrochloride and sodium acetate in 1:1 ratio3-Isobutyl-2,6-di(thiophe-3-yl)piperidin-4-one semicarbazone 4 and 3-Isobutyl-2,6-di(thiophe-3-yl)piperidin-4-one semicarbazone 5 were synthesized by Treatment of 3-isobutyl -2,6-di(thiophe-3-yl)piperidin-4-one1 with thiosemicarbazide and semicarbazide respectively .

Chemical data for compounds 1-5 are provided in Table 2.

3.1 CHARACTERIZATION COMPOUNDS

All synthesis compounds characterization in Table 1 .

TABLE 1: SPECTRAL DATA OF THE SYNTHESIZED COMPOUND

Compound	Spectral characterization data
1	<p>IR:cm⁻¹ :3309.02(N-H) , 2979.19,2929.49,2807.15,(C-H stretching) , 1700.82 (C=O) ,1455 .</p> <p>46 ,1294.86,1121.86,849.44(Ar-H) ,783.30,694.83.</p> <p>Mass:m/z : 319.9(M⁺),312.1,283.3,271.2,257.2,223.1,209.1, 206.0,195.3, 177.1,149.1,137.1,127.1,113.1,112.0,77.</p> <p>¹H-NMR:ppm:0.66,0.74(d,6H,2CH₃),1.42(m,2H,CH₂), 1.80(m,1H,CH) ,2.45(s,1H,NH),2.59(t,1H,CH-C=O) 2.68,2.81(d,2H,CH₂), 3.99(d,1H,CH-Ar), 4.38(t,1H,CH-Ar),6.93(t,CH-C-Ar),6.99(CH-C-Ar),7.23(CH-S),7.28(CH-S).</p> <p>¹³C-NMR:21.77(CH₃),23.43(CH₃),26.42(CH-(CH₃)₂),33.63(CH₂-C=O),52.22(C-3),56.70(C-5),57.12(C-2),62.98(C-6),123.83(C-2'),124.76(C-2'),</p>
2	<p>IR:cm⁻¹:2960.27(C-Hstretching),1705.06 (C=O) ,1628.01,1317.02,1266.74, 1077.13, 803.77(Ar-H),508.34.</p> <p>Mass m/z :549(M⁺),548, 547, 546, 545, 538, 537.6, 535.4, 500, 488,458,460, 393, 382, 367, 366, 278, 278, 264, 182, 100, 84,74,60.</p> <p>¹H-NMR:ppm:0.65,0.73(d,6H, (CH₃)₂) , 1.45(m,2H,CH₂), 1.69(m,1H,CH(CH₃)₂) 2.85(d,1H,CH₂-CH-CO) , 3.42,3.63 (m,2H,CH₂-CO),4.99(s,1H,-CH-NH), 5.22(s,1H,-CH-NH),7.12-7.70, 8.59(6H)- aromatic protons ,9.80 (s,2H, NH₂),10.20(Picryl ring proton).</p> <p>¹³C-NMR:(DMSO) :21.49(CH₃),23.12(CH₃),25.72(-CH-(CH₃)₂),33.67(CH₂- CH (CH₃)₂), 45.12(C-3), 51.12(C-5), 53.33(C-2),58.25(C-6) ,124.23, 125.19, 127.20, 127.95, 128.40, 128.40,128.86,129.79,141.81,160.78 (Aromatic</p>

	carbons), 201.78(C-4).
3	<p>IR:cm⁻¹:3309.02(N-H) ,3214.54(OH),2963.62,(C-H stretching) , 1242.12, 1076.14,994.77,695.75,418.</p> <p>Mass:m/z:335.1(M⁺),318,317,315,311,300,269,257, 244,234, 225</p> <p>,217,207,191,182,178, 163.2,153.2,100,84,74,60.</p> <p>¹H-NMR:CDCl₃:0.67,0.77(d,6H,(CH₃)₂),1.57,1.74(m, 2H,CH₂),</p> <p>2.06(t,2H,CH-(CH₃)₂) ,2.32(CH₂-CH-(CH₃)₂) ,2.46(t,1H,N=C-CH-CH₂)</p> <p>2.50(s,1H,NH),3.91(d,1H,CH-N),3.91(d,1H,CH-N),4.22(s,1H,N-OH),6.92-</p> <p>8.31(Aromatic protons).</p> <p>¹³C-NMR: CDCl₃:21.75(CH₃),23.77(CH₃),25.72(-CH(CH₃)₂), 34.99(CH₂-CH (CH₃)₂),35.06(C-3),48.22(C-5),56.33(C-2),63.84(C-6),123.77,124.39,125.08,125.68,126.16, 126.56:aromatic carbons,145.95(C-1'),147.24(C-1''),158.74(C-4).</p>
4	<p>IR:cm⁻¹ 3420.51(NH₂),3369.48(NH₂),3245.73,3148.64 ,2949.25,(C- stretching) ,1483.46(C=S),1386.17,1114.14(N-C-N),464.44.</p> <p>Mass:m/z 392(M⁺),378,376.1,355.1 ,338.3, 329.1 ,321 , 315, 300,297,282,266,265,209,206,193,182,167,163,154.2,84.2.</p> <p>¹H- NMR:CDCl₃:0.61,0.73(d,6H,(CH₃)₂),1.70,1.86(m, 2H,CH₂),</p> <p>=2.11(m,2H,CH-(CH₃)₂) ,2.36(CH₂-CH-(CH₃)₂) ,2.49(t,1H,N=C-CH-CH₂)</p> <p>3.87(d,1H,CH-N) ,4.36(d,1H,CH-N) , 6.98 (s,1H,NH) ,6.99-8.21(Aromatic protons),10.16(s,2H,NH₂)</p> <p>¹³C-NMR CDCl₃: 22.33(CH₃),24.24(CH₃) ,26.45CH(CH₃)₂, 35.33(CH₂-CH(CH₃)₂),38.09(C-3),50.63(C-5) ,56.63(C-2), 63.51(C-6) ,124.29,124.84,125.21,125.84,</p> <p>126.12,126.98,146.49,148.21,(Aromaticcarbons),154.66(C-4)179.73(C=S).</p>
5	<p>IR cm⁻¹: 3465.61(NH₂), 3183.42(NH),2952.34, 2867.21(C-H stretching),1693.30,1574.32(C=O),1434.05,696.07,596.23.</p> <p>Mass :m/z :376(M⁺),375.1,372.9,367.0,358.9,358.1,357.2,350. 8,349.7,345.7,341.9,336.1,334.0,332.9,331.8,</p> <p>300,297,282,266, 265, 209,206,193,182,167,163,154.2,84.2.</p> <p>¹H-NMR: CDCl₃: 0.64,0.82(d,6H,(CH₃)₂), 1.72,1.85(m, 2H,CH₂),2.28(m,2H,CH-(CH₃)₂), 2.45(CH₂-CH-(CH₃)₂) ,2.49(t,1H,N=C-CH-CH₂),4.20(s,1H,NH),4.39(d,1H,CH-N) ,6.03 (s,2H,NH₂),6.96(s,1H,NH), (6.99-8.46(Aromatic protons),</p> <p>¹³CNMR:CDCl₃:21.64(CH₃),23.34(CH₃),26.30CH(CH₃)₂,35. 25(CH₂- CH(CH₃)₂)</p> <p>37.11(C-3),49.67 (C-5) ,56.11(C-2),62.86(C-6) ,123.75,124.46 125.06,126.14,126.58,146.18, 146.56, 150.33(C=O), 158.04(C=N)</p>

TABLE 2: CHEMICAL DATA FOR COMPOUNDS (1-5)

Compound	R _f	m.p ⁰ C	Molecular formula	Yield%
1	0.95	120	C ₁₇ H ₂₁ NOS ₂	80.02%
2	0.86	185	C ₂₃ H ₂₄ N ₄ O ₈ S ₂	70.23%
3	0.52	163	C ₁₇ H ₂₄ N ₄ O ₈ S ₂	69.45%
4	0.96	202	C ₁₈ H ₂₄ N ₄ S ₃	63.45%
5	0.75	195	C ₁₈ H ₂₄ N ₄ OS ₂	45.23%

Scheme:1

3.2. Biological Activity

The synthesised compound (1-5) were tested for their antimicrobial activity in vitro against four species of bacteria (Klebsiella, Escherichia coli, Staphylococcus aureus, Salmonella typhi) using filter paper disc method. The screened compounds were dissolved individually in DMSO (dimethyl sulfoxide) in order to make up a solution of 1500-1000-500 ppm concentration for each of these compounds. Filter paper discs (Whatman No. 1 filter paper, 5 mm diameter) were saturated with the solution of these compounds. The discs were placed on the surface of solidified Nutrient agar dishes seeded by the tested bacteria. The diameters of inhibition zones (mm) were measured at the end of an incubation period, which was 24 h at 37°C for bacteria. Discs saturated with DMSO are used as solvent control. Streptomycin 25 ppm was used as reference substance for bacteria. Streptomycin was the first antibiotic used against Mycobacterium tuberculosis and it was used for years in monotherapy regimens, leading to the emergence of resistance. Besides its other antibiotic properties, it is one of the most widely used agents for dermatological reasons. In general, all the synthesized biolabile compounds 1-5 exhibited a wide

range of modest antibacterial activity especially against gram negative bacteria [14,15,16]. Tables 3 and 4.

TABLE 3: EFFECT OF THE COMPOUND (1,2) ON THE GROWTH OF BACTERIA (ZONE OF INHIBITION IN MM)

Micro organisms	Compound 3			Compound 4		
	500 ppm	1000 ppm	1500 ppm	500 ppm	1000 ppm	1500 ppm
Staphylococcus aureus	+	++	+++	-	+	++
Salmonella typhi	+	+	++	+	++	++
Escherichia coli	+	++	+++	+++	++	+++
Klebsiella	+	+	++	+	++	+++

(-)inactive, (+) weakly active (8-10), (+,+) moderately active (11-12), (+,+,+) strong active (13-14)

TABLE 4: EFFECT OF COMPOUND (3,4) ON THE GROWTH OF BACTERIA (ZONE OF INHIBITION IN MM)

Micro organisms	Compound 1			Compound 2		
	500 ppm	1000 ppm	1500 ppm	500 ppm	1000 ppm	1500 ppm
Staphylococcus aureus	-	+	++	-	+	++
Salmonella typhi	+	+	++	-	++	+++
Escherichia coli	+	++	+++	+++	++	+++
Klebsiella	+	++	+++	+	+++	+++

(-)inactive, (+) weakly active (8-10), (+,+) moderately active (11-12), (+,+,+) strong active (13-14)

4. CONCLUSION

Several substituted 3-Isobutyl-2,6-di(thiophen-3-yl) piperidin-4-one (1) and its derivatives 2-5 were synthesized. Compounds 1-5 were prepared, starting with thiophene-2-carbaldehyde, 5-methylhexan-2-one ketone, and ammonium acetate, via the Mannich reaction. Compounds 2 and 3 were synthesized via protonation and oximation. Compounds 4 and 5 were synthesized

with condensation with thiosemicarbazide and semicarbazide.¹H,¹³C-NMR Analysis show that this compounds were synthesis and this compound show good activity against grame positive bacteria more than grame negative bacteria.

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اصطناع ودراسة الفعالية الحيوية لمشتقات 3- ايزوبوتيل -6,2- ثنائي (تيوفنيل -3) بيبيريدون - 4

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الملخص:

ضمن إطار برنامج البحث عن الجديد في مجال المركبات المضادة للبكتيريا وللطفور تم في هذا البحث اصطناع مركب ايزوبوتيل - 6,2- ثنائي (تيوذفنيل -3) بيبيريدون - 4 و مشتقاته الأربع ووصفت هذه المركبات بالطرق المطيافية : $IR, MASS, ^1H-NMR, ^{13}C-NMR$.
اختبرت فعالية المركبات المصنعة ضد البكتيريا

Klebsiella pneumoniae, Escherichia coli, Staphylococcus aureus , Salmonella typhi

حيث استعمل streptomycin كمعيار.